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Key indicators

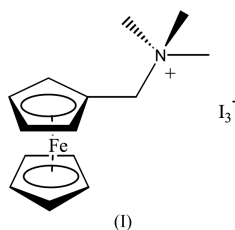
Single-crystal X-ray study
T = 150 K
Mean $\sigma(\text{C}-\text{C}) = 0.005 \text{ \AA}$
R factor = 0.026
wR factor = 0.062
Data-to-parameter ratio = 25.4For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

(Ferrocenylmethyl)trimethylammonium triiodide

The salt (ferrocenylmethyl)trimethylammonium iodide, $[\text{Fe}(\text{C}_5\text{H}_5)(\text{C}_9\text{H}_{15}\text{N})]\text{I}$, is used in an established procedure to attach ferrocenylmethyl frameworks to secondary amines. The title salt, $[\text{Fe}(\text{C}_5\text{H}_5)(\text{C}_9\text{H}_{15}\text{N})]\text{I}_3$, obtained by diffusion of Et_2O vapour into a CH_2Cl_2 solution containing equimolar $[\text{Fe}(\text{C}_5\text{H}_5)(\text{C}_9\text{H}_{15}\text{N})]\text{I}$ and I_2 , shows the expected cation balanced by an asymmetric but isolated triiodide counteranion [$\text{I}-\text{I} = 3.0006(4)$ and $2.8728(4) \text{ \AA}$].

Comment

A well established procedure for the introduction of ferrocenylmethyl frameworks into secondary amines involves the use of the salt (ferrocenylmethyl)trimethylammonium iodide, $[\text{Fe}(\text{C}_5\text{H}_5)(\text{C}_9\text{H}_{15}\text{N})]\text{I}$ (Beer *et al.*, 1991). A number of crystal structures containing the (ferrocenylmethyl)trimethylammonium cation, $[\text{Fe}(\text{C}_5\text{H}_5)(\text{C}_9\text{H}_{15}\text{N})]^+$, have been reported, including $[\text{Fe}(\text{C}_5\text{H}_5)(\text{C}_9\text{H}_{15}\text{N})]\text{I}$ (Ferguson *et al.*, 1994). The reaction of $[\text{Fe}(\text{C}_5\text{H}_5)(\text{C}_9\text{H}_{15}\text{N})]\text{I}$ (Lednicer & Hauser, 1960) with one equivalent of I_2 in CH_2Cl_2 gave the title compound, $[\text{Fe}(\text{C}_5\text{H}_5)(\text{C}_9\text{H}_{15}\text{N})]\text{I}_3$, (I), in good yield by diffusion of Et_2O vapour into the reaction mixture.



An X-ray diffraction analysis of (I) shows the expected cation balanced by an asymmetric triiodide counter-anion [$\text{I1}-\text{I2} = 3.0006(4)$ and $\text{I2}-\text{I3} = 2.8728(4) \text{ \AA}$]. Although the triiodide does not participate in any significant $\text{I} \cdots \text{I}$ intermolecular interactions, and the intramolecular geometry of the (ferrocenylmethyl)trimethylammonium cation shows no unusual features, there are a number of $\text{C}-\text{H} \cdots \text{I}$ contacts in the range $3.07-3.33 \text{ \AA}$. The shortest $\text{I} \cdots \text{C}$ contact here is one of $3.791(3) \text{ \AA}$ between atoms I3 and C15($1-x, 1-y, -z$), compared with one of $3.954(2) \text{ \AA}$ in $[\text{Fe}(\text{C}_5\text{H}_5)(\text{C}_9\text{H}_{15}\text{N})]\text{I}$ (Ferguson *et al.*, 1994).

Experimental

Et_2O vapour was allowed to diffuse into a CH_2Cl_2 (10 ml) solution of (ferrocenylmethyl)trimethylammonium iodide (0.05 g, 0.13 mmol) and I_2 (0.033 g, 0.13 mmol). Well shaped blocky crystals of the title

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compound were formed in 80% yield. Elemental analysis, found (calculated for $C_{14}H_{20}FeI_3N$): C 26.42 (26.32), H 3.24 (3.15), N 2.29 (2.20)%. FT Raman ($500\text{--}10\text{ cm}^{-1}$): 131.4, 101.1 cm^{-1} .

Crystal data

$[Fe(C_5H_5)(C_9H_{15}N)]I_3$
 $M_r = 638.86$
 Monoclinic, $P2_1/n$
 $a = 11.6959(9)\text{ \AA}$
 $b = 10.6136(9)\text{ \AA}$
 $c = 15.3474(12)\text{ \AA}$
 $\beta = 91.638(2)^\circ$
 $V = 1904.4(3)\text{ \AA}^3$
 $Z = 4$

$D_x = 2.228\text{ Mg m}^{-3}$
 Mo $K\alpha$ radiation
 Cell parameters from 7671 reflections
 $\theta = 2.3\text{--}27.6^\circ$
 $\mu = 5.64\text{ mm}^{-1}$
 $T = 150(2)\text{ K}$
 Block, red
 $0.50 \times 0.36 \times 0.30\text{ mm}$

Data collection

Bruker SMART APEX CCD area-detector diffractometer
 ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 2001)
 $T_{\min} = 0.064$, $T_{\max} = 0.180$
 13 480 measured reflections

4393 independent reflections
 3925 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.019$
 $\theta_{\text{max}} = 27.6^\circ$
 $h = -15 \rightarrow 14$
 $k = -13 \rightarrow 12$
 $l = -20 \rightarrow 19$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.026$
 $wR(F^2) = 0.062$
 $S = 1.09$
 4393 reflections
 173 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.029P)^2 + 2.356P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 1.08\text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -1.18\text{ e \AA}^{-3}$
 Extinction correction: SHELXL97 (Sheldrick, 1997)
 Extinction coefficient: 0.00106 (8)

Table 1

Selected geometric parameters (\AA , $^\circ$).

I1–I2	3.0006 (4)	I2–I3	2.8728 (4)
I1–I2–I3	178.600 (10)		

All H atoms were placed in geometrically calculated positions, with C–H distances of 0.98, 0.99 and 1.00 \AA for methyl, methylene and cyclopentadienyl groups, respectively, and with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. The largest residual electron-density peaks and holes all lie close to I atoms.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2002); data reduction: SAINT and SHELXTL (Bruker,

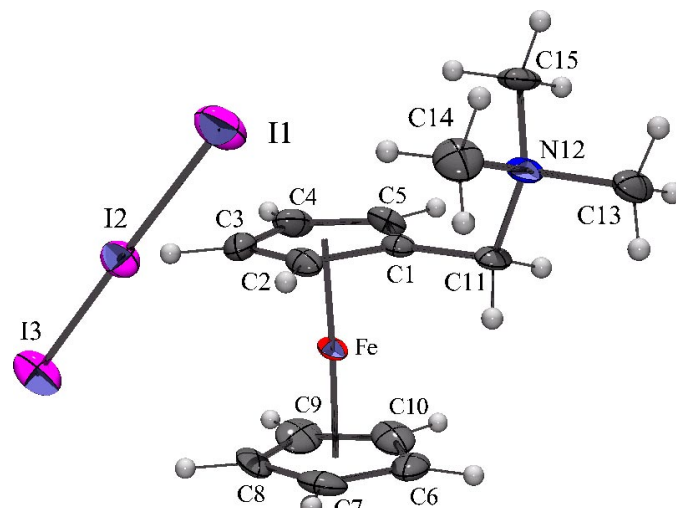


Figure 1

A view of the title salt, showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

2001); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: enCIFer (CCDC, 2003) and PLATON (Spek, 2003).

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